$^{\mathsf{IM}}$ national talent search examination (NTSE-2017) STAGE -1

TAMIL NADU STATE : SAT

Date: 6/11/2016

Max. Marks: 100 SOLUTIONS

Time allowed: 90 mins

101. When
$$10x^2 + x - 23$$
 is divided by $(2x + 3)$, the remainder is:

$$(2) - 2$$

Sol. On dividing
$$10x^2 + x - 23$$
 by $2x + 3$, remainder we get

Remainder =
$$10\left(\frac{-3}{2}\right)^2 - \frac{3}{2} - 23$$

$$=10\times\frac{9}{4}+\frac{-3-46}{2}=\frac{45}{2}-\frac{49}{2}$$

$$=\frac{-4}{2}=-2$$

102. If α and β are the zeros of the polynomial $25x^2 - 16$, then $\alpha^2 + \beta^2$ is:

(1)
$$\frac{32}{25}$$

(2)
$$\frac{25}{32}$$

(3)
$$\frac{25}{16}$$

(4)
$$\frac{16}{25}$$

Sol. Given polynomial
$$25 x^2 - 16 = 0$$

$$\alpha + \beta = 0, \, \alpha\beta = \frac{-16}{25}$$

$$(\alpha + \beta)^2 = \alpha^2 + \beta^2 + 2\alpha\beta$$

$$0-2\left(\frac{-16}{25}\right) = \alpha^2 + \beta^2 \implies \alpha^2 + \beta^2 = \frac{32}{25}$$

103. The sum of
$$\frac{a^3}{b-a}$$
 and $\frac{b^3}{a-b}$ is:

(1)
$$a^2 + ab + b^2$$

(1)
$$a^2 + ab + b^2$$
 (2) $-a^2 - ab - b^2$ (3) $a^2 - ab + b^2$ (4) $a^3 - b^3$

(3)
$$a^2 - ab + b^2$$

(4)
$$a^3 - b^3$$

Sol.
$$\frac{a^3}{b-a} + \frac{b^3}{a-b} = \frac{a^3}{b-a} - \frac{b^3}{b-a} = \frac{a^3-b^3}{b-a} = \frac{(a-b)(a^2+b^2+ab)}{(b-a)}$$

$$= -(a^2 + b^2 + ab) -a^2 - b^2 - ab$$

104.	Sum of the digits of a two digit number is 9. The number obtained by interchanging the digits is 18 more than twice
	the original number. The original number is:

(1) 72

(2) 27

(3) 36

(4) 63

Ans. (2)

Sol. Let the one's digit be x and ten's digit 9 - x

original number = x + 10(9 - x)

= 90 - 9x

Reversed number = 10x + 9 - x = 9x + 9

According to question

9x + 9 = 18 + 2(90 - 9x)

9x + 9 = 18 + 180 - 18x

27x = 189

n = 21/3 = 7

original number = $90 - 9 \times 7 = 90 - 63 = 27$

105. Which of the following are irrational numbers?

(i) $\sqrt{2+\sqrt{3}}$

(ii) $\sqrt{4+\sqrt{25}}$

(iii) $3\sqrt{5+\sqrt{7}}$

(iv) $\sqrt{6+3\sqrt{8}}$

(1) (i), (ii)

(2) (iii), (iv)

(3) (i), (iii)

(4) (i), (iv)

Ans. (NA)

Sol. (i) $\sqrt{2+\sqrt{3}}$ = irrational number

(ii) $\sqrt{4+\sqrt{25}} = \sqrt{4+5} = \sqrt{9} = 3$ = Rational number

(iii) $\sqrt[3]{5 + \sqrt{7}}$ = irrational

(iv) $\sqrt{6+3\sqrt{8}} = \sqrt{6+2} = \sqrt{8} = 2\sqrt{2}$ = irrational

Thus, (i), (iii), (iv) are irrational

106. For which value, point A (a, b) lies in the quadrant III:

(1) a > 0, b < 0

(2) a < 0, b < 0

(3) a > 0, b > 0 (4) a < 0, b > 0

Ans. (2)

Sol. For quadrant III a < 0, b < 0

107. If the LCM of 12 and 42 is (10 m + 4), then the value of 'm' is:

(1) 50

(2) 8

(3) $\frac{1}{5}$

(4) 1

Ans. (2)

Sol. LCM of 12, 42

 $LCM = 2 \times 3 \times 2 \times 7 = 84 = 10 \text{ m} + 4$

80 = 10 m

m = 8

- **108.** If the perimeter of a protactor is 72 cm then it's radius is $\left(take \pi = \frac{22}{7} \right)$:
 - (1) 7 cm
- (2) 21 cm
- (3) 14 cm
- (4) 3.5 cm

Ans. (3)

Sol.
$$\pi r + 2r = 72$$

$$r\left(\frac{22}{7} + 2\right) = 72$$

$$r \times \frac{36}{7} = 72$$

r = 14 cm

- **109.** The degree of the polynomial $(x + 1) (x^2 x x^4 + 1)$ is:
 - (1) 2

(2) 3

(3) 4

(4) 5

Ans. (4)

Sol.
$$(x + 1) (x^2 - x - x^4 + 1) = x^3 - x^2 - x^5 + x + x^2 - x - x^4 + 1$$

= $-x^5 - x^4 + x^3 + 1$.

- **110.** Two right circular cones have same radii. Ratio of their slant heights is 4 : 3. Then the ratio of their curved surface areas is :
 - (1) 16:9
- (2) 2:3
- (3) 4:3
- (4) 3:4

Ans. (3)

Sol. Ratio of curved surfaces =
$$\frac{\pi r \ell_1}{\pi r \ell_2} = \frac{4}{3}$$

- **111.** AB and CD are two chords of a circle which intersect each other externally at P. If AB = 4 cm, BP = 5 cm, PD = 3 cm, then the length of CD is:
 - (1) 10 cm
- (2) 12 cm
- (3) 8 cm
- (4) 11 cm

Ans. (2)

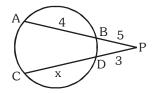
Sol. Let
$$CD = x$$

$$PA \times PB = PC \times PD$$

$$9 \times 5 = 3 \times (3 + x)$$

$$15 = 3 + x$$

$$x = 12$$



- **112.** The radii of two concentric circles are 7 cm and 14 cm respectively. The area between the two sectors of the circles whose central angle 60° is :
 - (1) 154 sq. cm
- (2) 77 sq. cm
- (3) 308 sq. cm
- (4) 98 sq. cm

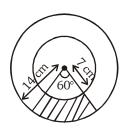
Ans. (2)

Sol. Area of shaded region

$$= \pi 14^2 \frac{60^\circ}{360^\circ} - \pi 7^2 \frac{60^\circ}{360^\circ}$$

$$=\pi \frac{60^{\circ}}{360^{\circ}} [14^2 - 7^2]$$

$$=\frac{22}{7} \times \frac{1}{6} (14 + 7) (14 - 7) = \frac{22 \times 2}{6} = 77 \text{ sq. cm}$$



113.	. Arithmetic mean of 20 observations is 15. If each observation is multiplied by $\frac{2}{3}$, then the arithmetic mean of them					
Ans.	is: (1) 10 (1)	(2) 30	(3) 45	(4) 15		
Sol.	New arithmetic mean =	$\frac{2}{3} \times 15 = 10$				
			One item is drawn at randor	m. The probabillity that it is non-		
Ans.	(1) $\frac{7}{10}$	(2) 0	(3) $\frac{3}{10}$	(4) $\frac{2}{3}$		
Sol.	Probability that the item of	drawn in non defective = $\frac{2}{1}$	$\frac{10-6}{10-6} = \frac{14}{10-10} = \frac{7}{10-10}$			
		f a circle has area equal to :	20 20 10			
	(1) $\frac{r^2}{2} \left(\frac{\pi}{2} - 1 \right)$ sq. units		(2) $\left(\frac{\pi}{4} + 1\right) r^2$ sq. units			
	(3) $\left(1-\frac{\pi}{2}\right)\frac{r^2}{2}$ sq. units		(4) $\left(\frac{\pi}{4}r^2 - 1\right)$ sq. units			
Ans. Sol.	(1) Segment of a quadrant or	f a circle has area				
	$= \frac{1}{4} \pi r^2 - \frac{1}{2} r^2$					
	$=\frac{r^2}{2}\left(\frac{1}{2}\pi-1\right)\text{sq. units}$					
116.	$\sqrt{1-\sin^2 A} \cdot \sqrt{\sec^2 A - 1}$					
Ans.	(1) 0 (3)	(2) 2	(3) 1	(4) –2		
Sol.	$\sqrt{1-\sin^2 A} \times \sqrt{\sec^2 A - 1}$	$\times \sqrt{1 + \cot^2 A}$				
	$= \cos A \times \tan A \times \csc C$ $= 1$	c A				
117.	If $5x = \csc 0$ and $\frac{5}{x} =$	$\cot \theta$ then $5\left(x^2 - \frac{1}{x^2}\right) =$				
	(1) 25	(2) 1	(3) $\frac{1}{5}$	(4) – 5		
Ans. Sol.		(1)				
	$\frac{5}{x} = \cot \theta$	(2)				
	squaring and subtracting	equation (1) & (2)				
	$25\left(x^2 - \frac{1}{x^2}\right) = \csc^2\theta$	$-\cot^2\theta \Rightarrow 25\left(x^2 - \frac{1}{x^2}\right) =$	1			
	$5\left(x^2 - \frac{1}{x^2}\right) = \frac{1}{5}$					

- **118.** If $x = a \cos \theta$, $y = a \sin \theta$, then $x^2 + y^2 =$

(3) a^2

(4) $a^2 + b^2$

- Ans. (3)
- **Sol.** $x = a \cos\theta$, $y = a \sin\theta$.

On squaring and adding

$$x^{2} + y^{2} = a^{2} \cos^{2}\theta + \sin^{2}\theta$$
$$= a^{2}.$$

- 119. If the diagonals of a rhombus are 30 cm and 40 cm, then the length of side of rhombus is : (1) 20 cm
 - (2) 22 cm
- (3) 25 cm
- (4) 45 cm

- Ans. (3)
- **Sol.** $d_1 = 30 \text{ cm} \Rightarrow \frac{d_1}{2} = 15 \text{ cm}$

$$d_2 = 40 \text{ cm} \Rightarrow \frac{d_2}{2} = 20 \text{ cm}$$

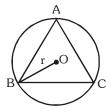
side =
$$\sqrt{\left(\frac{d_1}{2}\right)^2 + \left(\frac{d_2}{2}\right)^2}$$

$$= \sqrt{15^2 + 20^2}$$

= 25 cm



120. Equilateral triangle ABC is incribed in a circle. If side of the triangle = 24 cm, then the radius is _____.



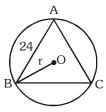
- (1) $6\sqrt{3}$ cm
- (2) $12\sqrt{3}$ cm
- (3) $8\sqrt{3}$ cm
- (4) 6 cm

- Ans. (3)
- Since in equilateral median and altitude coincides each other. Sol.

Altitude of the
$$\Delta = \frac{\sqrt{3}}{2} \times 24$$

$$= 12\sqrt{3}$$

Radius =
$$\frac{2}{3} \times 12\sqrt{3} = 8\sqrt{3}$$
 cm



- 121. Two cars A and B accelerate in the ratio of 2: 3 respectively. If they both accelerate for equal time, the ratio of their change in velocity is:
 - (1) 2:3
- $(2) \ 3:2$
- (3) 1:1
- (4) 1:2

- Ans. (1)
- **Sol.** $\frac{a_A}{a_B} = \frac{2}{3}, \frac{t_A}{t_B} = \frac{1}{1} \frac{V_A}{V_B} = ?$

$$a = \frac{v - u}{t}$$

$$a_A = \frac{v_A - u_A}{t_A} \ a_B = \frac{v_B - u_B}{t_B} \ \Rightarrow \frac{a_A}{a_B} = \frac{v_A - u_A}{v_B - u_B}$$

$$\therefore \quad \frac{v_A - u_A}{v_B - u_B} = \frac{2}{3}$$

- **122.** Two cars X and Y accelerate at the rate of 2m/s² and 3 m/s² respectively from rest. The ratio of time taken by the cars X and Y is 4:5. In that given ratio of time interval if the distance travelled by car X is 100 km then the distance travelled by car Y is:
 - (1) $\frac{1875}{8}$ km
- (2) $\frac{375}{2}$ km (3) $\frac{1875}{4}$ km
- (4) $\frac{375}{4}$ km

Sol. $a_x = 2m/s^2$, $a_y = 3 \text{ m/s}^2$, $u_x = u_y = 0$, $\frac{t_x}{t_y} = \frac{4}{5}$, $s_x = 100 \text{ km}$, $s_y = ?$

$$s = ut + \frac{1}{2} at^2$$

$$\frac{s_x}{s_y} = \frac{1/2a_x t_x^2}{1/2a_y b_y^2}$$

$$\frac{s_x}{s_y} = \frac{2}{3} \left(\frac{4}{5}\right)^2$$

$$\frac{100 \times 1000}{s_{_{U}}} = \frac{32}{75}$$

$$s_{y} = \frac{10^{5} \times 75}{32}$$

$$s_y = \frac{7500}{32} \text{ km} = \frac{1875}{8} \text{ km}$$

- 123. A car driver travelling with a uniform velocity of 2 m/s notices a railway level crossing at a distance of 435 m from him. And also he notices that it is going to be closed in 10 seconds. First he decides to cross the level crossing hence he accelerates his car at the rate of 2 ms⁻² for five seconds. Then he decides to stop the car. So he applies brakes and stops the car exactly before the level crossing (without following the timer). Calculate the minimum rate at which he has to decelerarte the car so that he stops the car exactly before the level crossing.
 - (1) $1.8 \,\mathrm{m/s^2}$
- (2) 18 m/s^2
- (3) 0.18 m/s^2
- (4) $3.6 \,\mathrm{m/s^2}$

Ans. (3)

Sol.

$$\begin{array}{c}
435 \text{ m} \\
A \quad a = 2\text{m/s}^2 \quad B \\
= 2\text{m/s} \quad t = 5 \text{ sec}
\end{array}$$

For the first five seconds

$$u = 2m/s$$
, $t = 5s$, $a = 2m/s^2$, $s = ? v = ?$

$$v = u + at$$

$$v = 2 + 2 \times 5 = 12 \text{ m/s}.$$

Now
$$v^2 = u^2 + 2as$$

$$144 = 4 + 2 \times 2 \times 5$$

$$s = \frac{140}{4} = 35 \text{ m}$$

From point B, remaining distance = 435 - 35 = 400 m

$$v = 0$$

$$u = 12 \text{ m/s}$$

$$\therefore v^2 = u^2 + 2as$$

$$0 = (12)^2 + 2 \times a \times 400$$

$$a = \frac{144}{800} = 0.18 \text{ m/s}^2.$$

- **124.** Two flies A and B revolve around a light in concentric circular path. The radius of circular path of A is twice of B. A travels with a uniform linear speed of 4 m/s while B travels with a uniform linear speed of 3 m/s. When A completes three full rounds then B would have completed
 - (1) 4 rounds
- (2) 3 rounds
- (3) 2 rounds
- (4) 1 round

Ans. (1)

$$\textit{Sol.} \quad r_{A}=2r_{B},\, v_{A}=4 \text{ m/s},\, v_{B}=3 \text{ m/s}$$

For A,
$$s_A = 2\pi(r \times 2)$$

$$v_A = 4 \text{ m/s}$$

$$t_A = \frac{s_A}{v_A} = \frac{4\pi r}{4m/s} = \pi r s$$

For three rounds $t = 3\pi r s$

For B

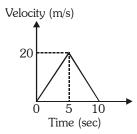
$$s_B = v_B \times t_B = 3 \times 3\pi r$$

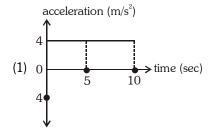
$$= 9 \pi r m$$

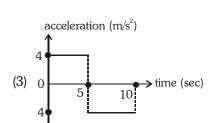
for one round = $s = 2\pi r$

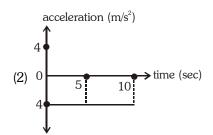
thus 4 round have been completed by fly B.

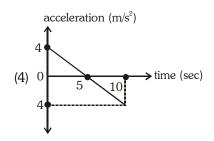
125. If the under given velocity (vs) time graph can be changed into acceleration (vs) time graph, then which one of the given options represents acceleration (vs) time graph:











Ans. (3)

Sol. Constant acceleration upto 5 sec and constant retardation for next 5 sec. is best reprsented by option 3.

- **126.** A boy travels along a circular path of radius 'r' m. When his angular displacement is $\frac{\pi}{3}$ radians then his linear displacement is:
 - (1) $r\sqrt{2} m$
- (2) r m
- (3) $2\sqrt{r}$ m
- (4) $\frac{\pi r}{3}$ m

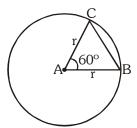
Ans. (2)

Sol. Angular displacement = $\frac{\pi}{3} = 60^{\circ}$

According to the figure, AB = AC = r

Thus $\angle ABC = \angle ACB = 60^{\circ}$

It is an equilateral triangle thus CB = r



- **127.** From a tower of height 20 m a boy throws a stone in the vertically upward direction with a velocity of 40 m/s and at the same time a girl drops another identical stone from the same tower. When the momentum of the stone dropped by the girl is maximum what will be displacement of the stone projected in the upward direction from the top of the tower? (take acceleration due to gravity of earth as 10 m/s²).
 - (1) 60 m
- (2) 40 m
- (3) 20 m
- (4) 0 m

Ans. (1)

Sol. For the stone dropped downward

$$u = 0$$
, $v = ?$, $g = 10 \text{m/s}^2$, $s = 20 \text{ m}$, $t = ?$

$$s = ut + \frac{1}{2} at^2$$

$$20 = \frac{1}{2} \times 10 \times t^2$$

t = 2s

for the stone thrown upward

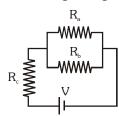
$$u = 40 \text{ m/s}, t = 2s, s = ?$$

$$s = ut + \frac{1}{2} gt^2$$

$$s = 40 \times 2 - \frac{1}{2} \times 10 \times 4$$

$$s = 80 - 20 = 60 \text{ m}$$

128. If all $R_a = R_b = R_c$ then the number of electrons travelling through R_a in every second is

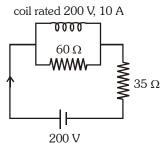


- (1) Half the number of electrons travelling through R_b
- (2) Equal to the number of electrons travelling through R_c
- (3) Twice the number of electrons travelling through R_c
- (4) Half the number of electrons travelling through R

Ans. (4)

Sol. The current which flows through Rc gets equally divided to R_a and R_b as they are equal. Thus, half the number of electrons travelling through R_c got to R_A and other half go to R_b .

129. The heat energy produced by the given coil in the given circuit in five minutes is



- (1) $6 \times 10^5 \,\mathrm{J}$
- (2) $5.4 \times 10^5 \,\mathrm{J}$
- (3) $6 \times 10^4 \,\text{J}$
- (4) $5.4 \times 10^4 \,\mathrm{J}$

Ans. (4)

Sol. Resistance of coil = $V/I = 20 \Omega$ Equivalent resistance of circuit =

$$\frac{1}{R_p} = \frac{1}{20} + \frac{1}{60} = \frac{1}{R_p} = \frac{4}{60} = R_p = 15 \Omega$$

Req. =
$$15 \Omega + 35 \Omega$$
. \Rightarrow Req = 50Ω

$$V = 200 \text{ V} \Rightarrow I = \frac{200}{50} = 4A$$

$$Current through coil = \frac{voltage \ drop}{R_{coil}}$$

voltage drop =
$$R_p \times I = 15 \times 4 = 60 \text{ V}$$

$$current through coil = \frac{60}{20} = 3A$$

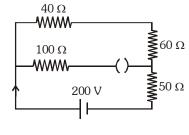
Heat produced = I^2 Rt

$$= 3 \times 3 \times 20 \times 5 \times 60$$

$$= 54000 \, J$$

$$= 5.4 \times 10^4 \,\mathrm{J}$$

130. The net current in the circuit is:

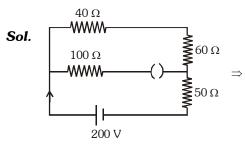


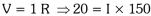
- (1) 2A
- (2) $\frac{4}{3}$ A
- (3) 1 A

50 Ω

(4) $\frac{2}{3}$ A

Ans. (2)





$$I = \frac{200}{150} = \frac{4}{3} A$$

200 V

100 Ω

www

131. A stone of mass 500 gm is dropped from a certain height. When it is exactly at the midpoint of its free fall, the kinetic energy possessed by it is 800 J. What is the height from where it is dropped?

(take acceleration due to gravity of earth as 10 ms⁻²)

- (1) 320 m
- (2) 160 m
- (3) 80 m
- (4) 240 m

Ans. (1)

Sol. Mass = $500 \, \text{gm}$

At mid point; KE = PE

$$KE = mg \frac{h}{2}$$

$$800 \text{ J} = \frac{500}{1000} \times 10 \times \frac{h}{2}$$

$$\frac{800 \times 2}{5} = h$$

$$h = 320 \text{ m}$$

- **132.** A car of mass 2,000 kg travelling with a uniform velocity of 2 m/s accelerates till its velocity becomes 22 m/s. The work done on the car is :
 - (1) 4.8 kJ
- (2) 480 kJ
- (3) 48 kJ
- (4) 500 kJ

Ans. (2)

Sol. m = 2000 kg, u = 2 m/s, v = 22 m/s

W. D =
$$\frac{1}{2}$$
 m (v² – u²)

$$=\frac{1}{2} \times 2000 (484 - 4) = 480 \text{ KJ}$$

- **133.** The engine of a bus of mass 5,000 kg accelerates the bus from 2 m/s to 20 m/s in 120 seconds. The power expended by the bus is :
 - (1) 8,250 W
- (2) 8.25 W
- (3) 82.5 W
- (4) 825 W

Ans. (1)

Sol. m = 5000 kg, u = 2 m/s, v = 20 m/s, t = 120 sec

$$P = \frac{W.D}{t} = \frac{1}{2} \text{ m} \frac{(v^2 - u^2)}{t} = \frac{1}{2} \times \frac{5000 (400 - 4)}{120} = 8250 \text{ W}.$$

- **134.** Tincture of iodine is a solution used as an antiseptic to clean wounds. This is prepared by dissolving solid iodine in:
- (1) alcohol
- (2) water
- (3) carbon di sulphide
- (4) ether

Ans. (1)

- **Sol.** Tincture of iodine is a solution of iodine in alcohol.
- **135.** You are provided with 64 g of sulphur in container A and 64 g of O_2 in container B. Which will have more number of molecules? (Atomic mass of S=32, O=16)
 - (1) 64 g of S

- (2) $64 \text{ g of } O_2$
- (3) both have equal number of molecules
- (4) cannot calculate with the given information.

Ans. (2)

Sol. Molecular mass of $O_2 = 32u$

Molecular mass of S or $\mathrm{S_8} = 32 \times 8 = 256~\mathrm{u}$

no. of molecules in 64g of $O_2 = \frac{64}{32} \times N_A = 2 N_A$

no. of molecules in 64g of S = $\frac{64}{256} \times N_A = \frac{1}{4} N_A$

$$2N_{\text{A}}>\frac{1}{4}N_{\text{A}}$$

 Shyam and Hari have 2 identical pieces of marble chips with same mass. They take equal volumes of dil. HCl with the same concentration in two different test tubes. Shyam puts the marble piece directly into the acid whereas Hari powdered the marble piece and puts it into the test tube. What will be the correct observation made? (1) Reaction in Shyam's test tube will be faster (3) Both reactions will happen in the same speed (4) No reaction happens in both the test tubes. Ans. (2) Sol. On powdering the marble pieces Hari increases the surface area of marble pieces which increases the rate of reaction. 137. Pi †apper is separately dipped into 2 different solutions X and Y. Colour of pH paper turned pale green in X and blue in Y. X and Y are most probably: (1) X – water, Y – NaOH (2) X – NaOH, Y – H₂O (3) X – HCl, Y – NaOH (4) X – NaOH, Y – HCl Ans. (2) Sol. p‡ paper is separately dipped into 2 different solutions and green colour in neutral solution. 138. An element has two shells and has double the number of electrons in its valence shell than the first shell. The valency of the element could be: (1) 8 (2) 4 (3) 2 (4) 6 Ans. (2) Sol. K shell consists maximum 2 electrons and its double is 4 that is present in valence shell. Electronic configuration of element = 2, 4 i.e. carbon it has valency – 4 139. Phiya and Karthick wanted to study about diffusion among liquids. They took identical beakers and poured 100 mL of H₂O in both the beakers. Priya heated the water to 50°C but Karthick maintained the water at room temperature. They both added 5 drops of ink into the beaker, what will they notice? (1) Colour of ink spreads faster in Karthick's beaker. (2) Colour of ink spreads faster in Karthick's beaker. (3) Colour of ink spreads faster in Fivals beaker. (4) In both the beakers, ink drops settle down at the bottom without spreading. Ans. (2) Sol. On heating the temperature energy of particles increases that's								
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 Ans. (1) Sol. On heating the temperature energy of particles increases that's why diffusion rate will be faster. 140. ²⁷/₁₃ Al looses electrons and froms trivalent cation. This ion will have (1) 13 electrons and 14 protons (2) 10 electrons and 13 protons (3) 10 electrons and 10 protons (4) 14 electrons and 13 protons Ans. (2) Sol. ²⁷/₁₃ Al ⇒ Atomic no. = no. of electron = no. of protons = 13. trivalent cation means it looses its 3 electrons. No. of electrons in ion = 10. 141. When CO₂ gas is passed through lime water, the solution turns milky. This is due to the formation of : 		(3) Colour of ink spreads a	at the same rate in both bea	kers.				
 Sol. On heating the temperature energy of particles increases that's why diffusion rate will be faster. 140. ²⁷₁₃ Al looses electrons and froms trivalent cation. This ion will have (1) 13 electrons and 14 protons (2) 10 electrons and 13 protons (3) 10 electrons and 10 protons (4) 14 electrons and 13 protons Ans. (2) Sol. ²⁷₁₃ Al ⇒ Atomic no. = no. of electron = no. of protons = 13. trivalent cation means it looses its 3 electrons. No. of electrons in ion = 10. 141. When CO₂ gas is passed through lime water, the solution turns milky. This is due to the formation of : 		(4) In both the beakers, ink drops settle down at the bottom without spreading.						
 140. ²⁷₁₃Al looses electrons and froms trivalent cation. This ion will have (1) 13 electrons and 14 protons (2) 10 electrons and 13 protons (3) 10 electrons and 10 protons (4) 14 electrons and 13 protons Ans. (2) Sol. ²⁷₁₃Al ⇒ Atomic no. = no. of electron = no. of protons = 13. trivalent cation means it looses its 3 electrons. No. of electrons in ion = 10. 141. When CO₂ gas is passed through lime water, the solution turns milky. This is due to the formation of : 	Ans.	(1)						
(1) 13 electrons and 14 protons (2) 10 electrons and 13 protons (3) 10 electrons and 10 protons (4) 14 electrons and 13 protons Ans. (2) Sol. ²⁷ ₁₃ Al ⇒ Atomic no. = no. of electron = no. of protons = 13. trivalent cation means it looses its 3 electrons. No. of electrons in ion = 10. 141. When CO ₂ gas is passed through lime water, the solution turns milky. This is due to the formation of :	Sol.	On heating the temperature	re energy of particles increa	ses that's why diffusion rate	will be faster.			
 (3) 10 electrons and 10 protons (4) 14 electrons and 13 protons Ans. (2) Sol. ²⁷₁₃Al ⇒ Atomic no. = no. of electron = no. of protons = 13. trivalent cation means it looses its 3 electrons. No. of electrons in ion = 10. 141. When CO₂ gas is passed through lime water, the solution turns milky. This is due to the formation of : 	140.	²⁷ ₁₃ Al looses electrons and	froms trivalent cation. This	ion will have				
 Ans. (2) Sol. 27/13 Al ⇒ Atomic no. = no. of electron = no. of protons = 13. trivalent cation means it looses its 3 electrons. No. of electrons in ion = 10. 141. When CO₂ gas is passed through lime water, the solution turns milky. This is due to the formation of : 		(1) 13 electrons and 14 pro	otons	(2) 10 electrons and 13 pr	rotons			
Sol. $^{27}_{13}\text{Al} \Rightarrow \text{Atomic no.} = \text{no. of electron} = \text{no. of protons} = 13.$ trivalent cation means it looses its 3 electrons. No. of electrons in ion = 10. 141. When CO_2 gas is passed through lime water, the solution turns milky. This is due to the formation of :		(3) 10 electrons and 10 pro	otons	(4) 14 electrons and 13 pr	rotons			
trivalent cation means it looses its 3 electrons. No. of electrons in ion = 10 . When CO_2 gas is passed through lime water, the solution turns milky. This is due to the formation of :	Ans.	(2)						
trivalent cation means it looses its 3 electrons. No. of electrons in ion = 10 . When CO_2 gas is passed through lime water, the solution turns milky. This is due to the formation of :	Sol.	$^{27}_{13}$ Al \Rightarrow Atomic no. = no.	. of electron $=$ no. of proto	ns = 13.				
No. of electrons in ion = 10 . 141. When CO_2 gas is passed through lime water, the solution turns milky. This is due to the formation of :		trivalent cation means it lo	poses its 3 electrons					
$\textbf{141.} \ \ \text{When CO}_2 \ \text{gas is passed through lime water, the solution turns milky. This is due to the formation of:}$								
-	141.			tion turns milkv. This is due	to the formation of :			
() 3/2 () 1/2		-						
Ans. (1)	Ans.	9	. ,	., . 3/2	, /2			
Sol. $CO_2 + Ca(OH)_2 \longrightarrow CaCO_3 + H_2O$		• •	CaCO ₂ + H ₀ O					
(lime water) (Milkiness)		-	ÿ 2					

- **142.** A set of students went on a nature trip where one of the students disturbed the honey comb, by throwing a stone on it. Few students were stung by the bee. A person gathering medicinal plants, came to their rescue and applied the extract of some leaves, which relieved the students of their pain. The chemical nature of leaf would have been:
 - (1) acidic
- (2) basic
- (3) neutral
- (4) mildly acidic

Ans. (2)

- **Sol.** Sting of bee contains formic acid which can be neutralise by basic solution.
- 143. Metal A reacts with water to give B. 'B' is used for white washing. On heating B gives C. C reacts with water to give back B. Identify A, B and C

Α

В

C

(1) Ca

CaO

Ca(OH)₂

(2) CaO

Ca

Ca(OH)₂

(3) Ca

Ca(OH)₂

CaO

(4) CaO

Ca(OH)₂

Ca

Ans. (3)

Sol. (A) (B) (C) (C) $Ca + H_2O \rightarrow \underbrace{Ca(OH)_2}_{used \ as} + H_2 \qquad Ca(OH)_2 \xrightarrow{\Delta} CaO + H_2O \xrightarrow{Ca(OH)_2} Ca(OH)_2$

144. P, Q and R are 3 metals that undergo chemical reactions as follows:

$$P_2O3 + 2Q \rightarrow Q_2O_3 + 2P$$

$$2P + 3RO \rightarrow P_2O_3 + 3R$$

$$2RSO_4 + 2Q \rightarrow Q2(SO_4)_3 + 2R$$

observe the reactions and arrange the metals in the increasing order of their reactivity.

- (1) R, P, Q
- (2) Q, P, R
- (3) P.Q.R
- (4) Q, R, P

Ans. (1)

Sol. Q can replace P from P_2O_3

Q is more reactive than P

P can replace R from RO

P is more reactive than R

Q > P > R

- **145.** Which among the following is the correct representation of $360 \, \mathrm{g}$ of water (H=1, O = 16)
 - (i) 2 moles

(ii) 20 moles

(iii) 6.022×10^{23} molecules

(iv) 1.2044×10^{25} molecules

- (1) (i) and (iii)
- (2) (ii) and (iv)
- (3) (i) and (iv)
- (4) (ii) and (iii)

Ans. (2)

Sol. no. of moles $\Rightarrow \frac{360}{18} 20$ moles

no. of molecules = $20 \times 6.022 \times 10^{23} = 1.2044 \times 10$ molecules

- 146. Metallic copper can be used to retrieve silver from silver nitrate solution. This is because
 - (1) Cu is less reactive than Ag

- (2) Cu is more reactive than Ag
- (3) Cu and Ag have same reactivity
- (4) Cu doesn't react with AgNO₃

Ans. (2)

Sol. Cu is more reactive than silver that's why it can replace Ag from silver nitrate.

147.
$$6CO_2 + \xrightarrow{?} \xrightarrow{Sunlight} ?$$

$$C_6H_{12}O_6+6O_2+6H_2O$$

which two raw materials required for photosynthesis are missing in the above equation?

(1) Oxygen and Water

(2) Oxygen and Calcium

(3) Water and Chlorophyll

(4) Chlorophyll and Oxygen

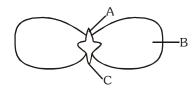
Ans. (3)

$$\textbf{Sol.} \quad 6\text{CO}_2 + 12\text{H}_2\text{O} \xrightarrow{\quad \text{Sunlight} \\ \quad \text{Chlorophyll} } \quad \text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2 + 6\text{H}_2\text{O}$$

- 148. In a Bamboo plant, the water reaches all the parts of the plant. Name the force that helps in this process:
 - (1) Diffusion
- (2) Transpirational pull
- (3) Gravitational pull
- (4) Translocation

Ans. (2)

- **Sol.** Transpirational pull is the major force which helps in ascent of sap i.e. water and minerals in tall trees.
- **149.** Choose the correct arrangement of the parts A, B and C marked in the given figure.



- (1) Cotyledon, Plumule and Radicle
- (2) Plumule, Cotyledon and Radicle
- (3) Radicle, Plumule and Cotyledon
- (4) Radicle, Cotyledon and Plumule

Ans. (2)

- **Sol.** A is plumule, B is cotyledon and C is radicle.
- **150.** The production of orchids by the method of Tissue Culture is also known as:
 - (1) Vegetative propagation

(2) Micro propagation

(3) Fragmentation

(4) Regeneration

Ans. (2)

- **Sol.** Tissue culture is also known as micropropagation.
- **151.** If a nail is hammered into the tree trunk, then the position of the nail after few years will be:
 - (1) same
- (2) above
- (3) lower
- (4) nail will disappear

Ans. (1)

- **Sol.** The nail will remain at same position after few years. This is because a plant grows at its tip (stem or root) not from the point at which it joins the ground.
- **152.** Which one of the following is the correct Hierarchy of classification?

(1)	Kingdom
	Division
	Class
	Order
	Family
	Genus
	Species

(2) Kingdom
Division
Order
Class
Family
Genus
Species

(3) Kingdom
Division
Class
Order
Genus
Family
Species

(4) Kingdom
Division
Class
Order
Family
Species
Genus

Ans. (1)

Sol. The correct Hierarchy of classification is

Kingdom; phylum/division; class; order; family; genus; species

153 .	What will happen to	the cell, if the medium has	s a lower concentration of water	than the cell?					
	(1) bulge	(2) shrink	(3) no change	(4) cannot be predicted					
Ans.	(2)								
Sol.	The cell kept in a me	dium having a lower conce	entration of water will shrink due	to exosmosis.					
154 .	Assertion (A) : Peo	Assertion (A): People entering into the burning place die due to suffocation.							
	Reason (R): Smoke	Reason (R) : Smoke contains large amount of carbon mono oxide, a toxic gas.							
	(1) (A) is correct and	l (R) is wrong							
	(2) (R) explains (A)								
	(3) (R) does not expl	lain (A)							
	(4) (A) is wrong but ((R) is correct							
Ans.	(2)								
Sol.	-		th due to suffocation as haemogl obin from carrying oxygen to tissu	obin has more affinity for carbon ues thus causing suffocation.					
155 .	Study the relationshi	p of the given pairs and ch	oose the correct option to fill in t	he blank.					
	Oestrogen: Oogensis	3							
	Prolactin: Lactation								
	Oxytocin:	-							
	(1) Thickness of endo	ometrium							
	(2) Secondary sexual characters								
	(3) Rhythmic contract	ction of uterus during deliv	ery of the baby						
	(4) Provides protecti	on against intestinal and re	espiratory functions						
Ans.	(3)								
Sol.	Oxytocin causes rhyt	hmic contraction of uterus	during delivery of baby.						
156 .	The carcinogenic tox	ic gas released during ciga	rette smoking is :						
	(1) Nitrogen oxide		(2) Methyl Iso cyanate						
	(3) Methyl mercury		(4) Benzopyrene						
Ans.	(4)								
Sol.	Benzopyrene is a care	cinogenic toxic gas released	d during cigarette smoking.						
<i>157.</i>		_	ch part of the brain is affected?						
	(1) Cerebrum	(2) Cerebellum	(3) Medulla oblongata	(4) Hypothalamus					
Ans.	, ,								
Sol.	Part of the brain resp and precision of body	_	lance and posture is cerebellum. I	It also contributes to coordination					
158 .	In case of snake bite,	, doctor treats the patient, v	with preformed antibodies. What	t type of immunity it develops?					
	(1) Innate immunity								
	(2) Naturally Active A	Acquired immunity							
	(3) Artificially Active	Acquired immunity							
	(4) Naturally Passive	Acquired immunity							
Ans.	(NA)								
Sol.	The immunity which	develops when the doctor t	treats with preformed antibodies v	vill be Artificially Acquired Passive					

immunity which is not in option.

159. Match the organisms given in Column-I with the nutritional processes given in Column-II.

Column-I	Column-II
(a) Leech	(i) Holozoic Nutrition
(b) Amoeba	(ii) Autotrophic Nutrition
(c) Mushroom	(iii) Parasitic Nutrition
(d) Green plant	(iv) Saprophytic Nutrition

	(a)	(b)	(c)	(d)
(1)	(ii)	(iv)	(i)	(iii)
(2)	(iii)	(i)	(iv)	(ii)
(3)	(i)	(iv)	(iii)	(ii)
(4)	(iv)	(iii)	(ii)	(i)

Ans. (2)

- **Sol.** Leech shows parasitic nutrition. Amoeba has holozoic nutrition. Mushroom is a saprophyte which feeds on dead and decaying organisms. Green plant is an autotroph i.e. it makes its own food.
- **160.** Mendel crossed tall plant with dwarf plant in his famous experiment on *Pisum sativum*. In the first generation, he got only tall plants. Because:
 - (1) The parental plants were heterogenous to their characters
 - (2) The soil was fertile
 - (3) The parental plants were pure to their characters
 - (4) The tallness character was a recessive character

Ans. (3)

Sol. As the parental plants were pure to their characters i.e. they were homozygous tall and dwarf so he got only tall plants in F_1 generation.

$$TT \times tt$$

Tt Tt $(F_1 generation)$

161. The treaty concluded after the II Indo-China war was_____.

(1)	Treaty of Nanking
1 1	field of Franking

(2) Treaty of Peking

(3) Treaty of Shimonoseki (4) Treaty of London

Ans. (1)

Sol. Treaty of Nanking was signed after the Second India China war which is also known as the Second Opium War.

162. Mussolini was the editor of Socialist Newspaper called ______.

(1) New India

(2) Avanti

(3) Mein Kamph

(4) Social Contract

Ans. (2)

Sol. Mussolini edited Avanti

163. The working languages of the United Nations are

(1) Arabic and Chinese

(2) Chinese and English

(3) English and French

(4) Russian and Spanish

Ans. (3)

Sol. In the Economic and Social Council, as of 1992, there are six official languages (Arabic, Chinese, English, French, Russian and Spanish) of which three are working languages (English, French, and Spanish). Later, Arabic, Chinese, and Russian were added as working languages in the Economic and Social Council.

164. The Indian who headed the United Nations General Assembly in 1953 was

(1) Mrs. Vijayalakshmi Pandit

(2) Moovalur Ramamirdham Ammaiyar

(3) Dr. Muthulakshmi Reddy

(4) Dr. S. Dharmambal

Ans. (1)

Sol. Between 1946 and 1968, Mrs. Vijayalakshmi Pandit headed the Indian delegation to the United Nations. In 1953, she became the first woman President of the United Nations General Assembly.

<i>165</i> .	Pick the odd one out: N	ck the odd one out : Neelakesi, Choolamani, Yapperumkalam, Kundalakesi.						
	(1) Choolamani	(2) Kundalakesi	(3) Neelakesi	(4) Yapperumkalam				
Ans.	(2)							
Sol.	Kundalakesi is a fragme	entary Tamil Buddhist epic	written by Nathakuthanaar	. All others are related to Jainism.				
166.	was known	as the 'World's First Com	piler of Law'.					
	(1) Napoleon III	(2) Hammurabi	(3) Confucius	(4) Cheops Khufu				
Ans.	(2)							
Sol.	Hammurabi is consider	ed as world's first compiler	of laws.					
167.	Plato wrote	<u>_</u> ·						
	(1) The Republic		(2) The Law of Twelve	Tables				
	(3) Justinian Code		(4) Meditations					
Ans.	(1)							
Sol.	The Republic is a Socra	tic dialogue, written by Pla	nto around 380 BC.					
168.	was defe	ated in the Battle of Water	rloo.					
	(1) Hitler	(2) Mussolini	(3) Stalin	(4) Napoleon Bonaparte				
Ans.	(1)							
Sol.	Hitler was defeated in the	he Battle of Waterloo in 18	315.					
169.	The Brahadeeswarar te	mple was built by the	·					
	(1) Cheras	(2) Pandyas	(3) Pallavas	(4) Cholas				
Ans.	(4)							
Sol.		CE by Raja Raja Chola in ars old in September 2010.		emple popularly known as the 'Big				
170 .	The tower temples were	e also known as	_·					
	(1) Ziggurats	(2) Pyramids	(3) Hanging Garden	(4) Tower of Babel				
Ans.	(1)							
Sol.	Tower temples were also	known as Ziggurats						
171.	'Man is the maker of his	own destiny' was stressed	by					
	(1) Gauthama	(2) Mahavira	(3) Laotze	(4) Zoroaster				
Ans.	(Bonus)							
Sol.	This line was coined by	Swami Vivekanand.						
172 .	The Longitude that help	os us to calculate the India	n Standard Time is :					
	(1) 80° E	(2) 82° 30' E	(3) 82°50' E	(4) 81° E				
Ans.	(2)							
Sol.	It is taken as the standard meridian of India which helps in calculating the Indian Standard Time.							
173.	'The Sorrow of Bihar' is	:						
	(1) Kosi	(2) Yamuna	(3) Brahmaputra	(4) Ganga				
Ans.	(1)							
Sol.	Kosi is the sorrow of Bih	ıar.						
174.	There is enough for ever	rybody's need and not for a	anybody's greed.' It was voic	ed out by :				
	(1) Mahatma Gandhi	(2) Jawaharlal Nehru	(3) Medha Patkar	(4) Indira Gandhi				
Ans.	(1)							
Sol.	Mahatma stated this sta	atement .						

175.	The main objective of Na	ational Forest Policy is to :				
	(1) bring 33% of geograp	hical area under forests.				
	(2) bring 20% of geograp	hical area under forests.				
	(3) maintain 30% of geog	graphical area under forests	5.			
	(4) bring 35% of geographical area under forests.					
Ans.	(1)					
Sol.	Indian Forest Policy (Evolution Land area of the country Count	_	ement to Community stre	ess on having at least 33 percent of		
176 .	Compressed Natural Gas	(CNG) is becoming more	popular because :			
	(1) Available at cheaper i	ate				
	(2) Low emission of carbo	on dioxide				
	(3) It is used in power and	d fertilizer Industries				
	(4) None of the above					
Ans.	(2)					
Sol.	CNG is environment frier	ndly fuel and hence is beco	ming more popular.			
<i>177</i> .	Choose the correct order	of arrangements, the type	s of coal according to its qu	ality/ carbon content.		
	(1) Anthracite, Bitumino	us, Lignite, Charcoal.				
	(2) Anthracite, Charcoal,	Bituminous, Lignite.				
	(3) Anthracite, Lignite, (Charcoal, Bituminous.				
	(4) Bituminous, Lignite, A	Anthracite, Charcoal.				
Ans.	(1)					
Sol.	As per the given sequence	e Anthracite has largest car	bon content followed by of	her three.		
178.	The first state in India wh the state :	ich has made roof top rain	water harvesting structure o	compulsory to all the houses across		
	(1) Rajasthan	(2) Maharashtra	(3) Karnataka	(4) Tamil Nadu		
Ans.	(4)					
Sol.	Tamil Nadu has made roo	of top rainwater harvesting	compulsory.			
179.	Srirangam is a/an					
	(1) Island	(2) Plateau	(3) Coastal Plain	(4) Hilly		
Ans.	(1)					
Sol.	Srirangam (formerly Vel Tiruchirapalli, in South Ir		niruvarangam in Tamil) is a	an island and a part of the city of		
180.	Shrinking of forest cover	is mainly because of :				
	(1) Over population	(2) Urbanization	(3) Industrialization	(4) Farming activities		
Ans.	(1)					
Sol.	An increase in tribal popu	ulation has resulted in loss	of forest cover			
181.	'Finland of Tamilnadu' is	:				
	(1) Kancheepuram	(2) Villupuram	(3) Ooty	(4) Tirunelveli		
Ans.	(1)					
Sol.	Kancheepuram district is	called Finland of Tamil Na	du			
182.	Geographical surname, "	Detroit of Southern Asia" r	efers to :			
	(1) Hengaluru	(2) Mumbai	(3) Chennai	(4) New Delhi		
Ans.	(3)					
Sol.		g "manufacturing" base for Chennai the Detroit of Sou		of the manufacturers have 40 years		

183.	8. Which Article of our Constitution prohibits any child below the age of 14 from working in dangerous, hazardo employment like mining?					ıs	
	(1) 19	(3)	24	(2) 23	(4) 26		
Ans.		(3)	2 4	(2) 23	(4) 20		
Sol.	` '	tion of India in th	a Fundamar	ital Rights and the Directive I	Principles of State Policy prohibits chi	14	
301.				=	ed in any other hazardous employmen		
184.	Name the Pre	esidency Constitue	ency in which	women were enfranchised fo	r the first time in India.		
	(1) Madras	(2)	Bombay	(3) Calcutta	(4) Bengal		
Ans.	(1)						
Sol.	Women were	first elected in Ma	dras presider	ncy			
185 .	The Presiden	t of World Bank is	always the c	itizen of :			
	(1) UK	(2)	USA	(3) Russia	(4) France		
Ans.	(2)						
Sol.	The presiden	t of World Bank h	as always bee	en from USA			
186.	Name the co	untry which has si	ngle party sys	etem.			
	(1) China	(2)	Britain	(3) Singapore	(4) Ghana		
Ans.	(1)						
Sol.	China has a s	single party system	n, Communis	t Party rules China			
187.	Which Indian	n state has its own	Constitution	?			
	(1) Jammu a	nd Kashmir (2)	Maharashtra	(3) Uttarakhand	(4) Nagaland		
Ans.	(1)						
Sol.	Jammu and l	Kashmir has its ov	n Constitutio	on, which is explained in Artic	le 370 of the Indian Constitution.		
188.	Here are som	ne of the guiding v	alues of the C	Constitution and their meaning	gs. Match them correctly :		
		Guid	ina	Meanin	, a		
		(a) Sovereign		Government will not favour			
		(b) Republic	(ii)	People have the supreme rig			
		(c) Fraternity	(iii)	The state is ruled by the elec	The state is ruled by the elected representatives.		
		(d) Secular	(iv)	People should live like broth	ers and sisters.		
	(1)	(2)	(3)	(4)			
	(1) (ii)	(i)	(iv)	(iii)			
	(2) (iii)	(iv)	(iv)	(iii)			
	(3) (iii)	(iii)	(iv)	(iii)			
	(4) (iv)	(iii)	(iv)	(iii)			
Ans.	(4)						
Sol.	-	en sequence, opt					
189.	What is the ro	ole of 'Amnesty In	ternational'?				

- $(1) \ To \ work \ for \ International \ peace$
- (2) To stop arms race in the world
- (3) Collecting information about condition of international prisoners
- (4) None of the above

Ans.	(4)						
Sol.	Amnesty International is a Human Rights Organisation, and aims at securing human rights.						
190.	is rightly known as the 'Guardian of the Constitution'.						
	(1) District Court	(2) Magistrate Court	(3) High Court	(4) Supreme Court			
Ans.	(4)						
Sol.	Supreme Court is the Gu	uardian or the Custodian of	the Constitution.				
191.	Which state has bicamen	al legislatures ?					
	(1) Tamil Nadu	(2) Gujrat	(3) Bihar	(4) Kerala			
Ans.	(3)						
Sol.		ndhra Pradesh, Telangana, I Legislatures, these are call		Karnataka, Maharashtra and Uttar Ihan Parishad)			
192 .	Name the Chief Election	Commissioner of India.					
	(1) Nasim Zaidi	(2) Rajesh Lakhoni	(3) H.S. Brahma	(4) V.S. Sampath			
Ans.	(1)						
Sol.		awat. H.S.Brahma retired o		on Commissioners are Achal Kumar hen Dr.Nasim Zaidi took over as 20th			
193.	Pick the odd man out:						
	(1) Mrs. Sumitra Mahaja	n	(2) Mrs. Sushma Swa	raj			
	(3) Mrs. Meera Kumar		(4) Mrs. Najma Heptu	ıllah			
Ans.	(2)						
Sol.	Mrs. Sushma Swaraj has	s never been the Speaker o	f Lok Sabha or the chairn	nan of Rajya Sabha.			
194.	The growth rate of a cou	ıntry is decided by	_·				
	(1) The growth in literacy	<i>y</i> rate	(2) The growth in er	mployment opportunities			
	(3) The quality of the po	pulation	(4) The growth of the	economy			
Ans.	(4)						
Sol.	The growth rate of a cou	ntry is determined by the g	rowth of the economy of t	that country.			
195.	The state that has the lo	west Infant Mortality Rate	in India is :				
	(1) Andhra Pradesh	(2) Tamil Nadu	(3) Kerala	(4) Rajasthan			
Ans.	(3)						
Sol.	-	-		Kerala with 12 deaths per 1,000 live ed by Uttar Pradesh and Odisha with			
196 .	The Head of the Plannir	ng Commission in India is :					
	(1) The Vice President	(2) The Prime Minister	(3) The President	(4) A Cabinet Minister			
Ans.	(2)						
Sol.	The Prime Minister is the	e head of the Planning Con	nmission of India.				
197.	Pick the odd one out :						
	(1) Income Tax	(2) Road Tax	(3) Water Tax	(4) Property Tax			
Ans.	(1)						
Sol.	All other taxes are applied on public facilities.						

198.	The process of withdrawal of United Kingdom from the European Union is called:			
	(1) BREXIT	(2) BRIXTON	(3) BRICS	(4) BREXTON
Ans.	(1)			
Sol.	BREXIT - Britain's Exit from the European Union.			
199.	If a mother is taking care of children and household activities within the walls of the house, what kind of activity is she performing?			
	(1) Market activity	(2) Non - market activity	(3) Economic activity	(4) Non - economic activity
Ans.	(2)			
Sol.	A mother taking care of his children is a non economic activity.			
200 .	'Green Revolution' is associated with the production of :			
	(1) Sugar	(2) Pulses	(3) Wheat	(4) Cereals
Ans.	(3)			
Sol.	Green Revolution is associated with wheat production which was later extended to Rice.			